

reaction. Conversely, the addition of molecular oxygen increases the ethylene quantum yield from 0.14 to 0.38 at a total pressure of 1 Torr.

From these results it is not obvious what mechanism is responsible for such an effect. However, the involvement of excited *triplet* intermediates may be assumed:



Direct absorption of a photon leads to the formation of a *singlet* excited molecule. This excited molecule decomposes into two *triplet* excited molecules etc. Triplet–triplet annihilation processes are well known in the liquid phase but are not often included in gaseous chemical systems.

Collision-induced radiationless transitions in CS₂

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We measured the quenching rate constants of CS₂ vapour (¹A₂ and ³A₂ states) by laser-induced fluorescence with the gases CS₂, CO₂, O₂, NO, C₂H₅OH, CH₃OH, C₂H₅NO₂, CCl₄ and CH₃CN. The emission was measured in regions centred at 4480 and 5860 Å. The quenching cross sections were obtained for both electronic states. Stern–Volmer plots at 4480 Å for the ³A₂ state show saturation effects for CH₃CN, C₂H₅NO₂ and O₂, whereas only O₂ produces such effects at 5860 Å. The quenching cross sections correlate with the molecular parameters defined in the Thayer–Yardley model for collision-induced radiationless transitions. The decay of CS₂ obeys this model for the two states involved.

Energy selection experiments in glassy matrixes: the fluorescence spectrum of isobacteriochlorin

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The fluorescence spectrum of a synthetic isobacteriochlorin 1 was studied in a glassy matrix at low temperature. It consists of a single strong 0–0 band and very little vibronic structure. Excitation into a region 400 - 900 cm⁻¹ above the